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By Evan P. Silberstein Elements are simple things That can't be broken down. When put together chemically, They always form compounds. But thinking of the usual way That chemicals are found, It's jumbled up in mixtures that Most chemicals abound.

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Answer the questions below based on your reading above and on your knowledge of chemistry. Answer questions 1-4 by referring to the diagram to the right showing an electrochemical cell. The metal at electrode A is silver. The metal at electrode C is lead. The electrolytes at locations B, D, and E are potassium nitrate, silver nitrate,

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Sample Problem 1 What is the concentration of a 30. mL sample of HCl if it can be neutralized by 50. mL of 1.2 M NaOH? Step 1: Note the ratio of H⁺ to OH⁻ is 1 to 1 Step 2: Substitute values into the equation $M_A \times V_A = M_B \times V_B$ $M_A(30. \text{ mL}) = (1.2 \text{ M})(50. \text{ mL})$ Step 3: Solve for the unknown M 12M 50mL

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