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9: Chemical Reactions.

9.1: Writing chemical reactions. 9.2

Balancing Reactions.

9.3: Reaction Types. ...

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Handouts, Reviews,

and Answer Keys.

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Unit 1 - Intro to
Chemistry. Unit 4 -
Electrons. Unit 2 -
Matter and Energy.
Unit 3 - Atoms and
Moles. Unit 5 - Periodic
Table. Unit 6 - Ionic
Compounds. Unit 7 -
Formulas. Semester 2.
Unit 8 - Covalent
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Chemical Reactions.
Unit 10 -
Stoichiometry. Unit 11
- Gases. Unit 12 -

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Structure, Unit 3
Electron, Unit 4
periodic table, Unit 5:
bonding, Unit
6:chemical
nomenclature, Unit
7:the mole, Unit
8:chemical reactions
Terms in this set (126)

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**Study 126 Terms |
Chemistry semester
1 course review ...**

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Chemistry Textbook:
CHAPTER SUMMARIES
(Chapter 1-5)

Chemistry Vocabulary
Words: CHEMISTRY

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Notes List: Notes List;

Sample Exams (from
previous years) Sample

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Part 1: Ions

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Questions (see old

chapter tests) (~30)

Sample Part 3:

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Unit 4; Unit 5;

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Stoichiometry,

Chemical Reactions,

and First Semester

ReviewUnit 5: Unit 6:

EnergyUnit 6: Unit 7:

Earth's

AtmosphereUnit 7: Unit

8: Water QualityUnit 8:

Unit 9: Reaction Rates

and EquilibriumUnit 9:

Unit 10: Nuclear

Chemistry and Final

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Questions |

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Semester 1 Unit 9

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Semester I Exam

Review Questions 1. 2.

3. John Dalton, Ernest Rutherford, and J.J.

Thomson all

contributed to our understanding of the

atom. Choose one of the scientists listed

above and explain his contribution to our

current understanding of atomic theory and/

or atomic models. You need to include

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